



1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

Product Name MANGANESE SULPHATE MONOHYDRATE

Other Names Manganese Sulphate Manganous(ii) Sulphate, Hydrate; Manganese Sulphate Monohydrate

Uses Fertilisers, feed additive, paints and varnishes, ceramics, textile dyes, medicines, fungicides, ore

flotation, catalyst in viscose process, synthetic manganese dioxide, pharmaceutical applications.

Chemical FamilyNo Data AvailableChemical FormulaMnSO4.H20

Chemical Name Manganese Sulphate Monohydrate

Product Description No Data Available

Contact Information Australia Location Telephone Ask For

Rural Liquid Fertilisers Pty Ltd 61 Dowd Street +61 1800 753 000 Technical Officer

Welshpool WA 6106

2. HAZARDS IDENTIFICATION

ADG Code Non-Dangerous Goods according to the criteria of the Australian Dangerous Goods

Code (ADG Code).

ASCC Hazardous Classification Hazardous according to the criteria of ASCC [NOHSC:1008(2004)]

Categories Xn Harmful

N Dangerous For The Environment

Risk Phrases R48/20/22 Harmful : danger of serious damage to health by prolonged exposure through

inhalation and if swallowed.

R51/53 Toxic to aquatic organisms; may cause long term adverse effects in the

aquatic environment.

Safety Phrases S22 Do not breathe dust.

S36 Wear suitable protective clothing.S51 Use only in well ventilated areas.

S53 Avoid exposure - obtain special instructions before use.

S61 Avoid release to the environment. Refer to special instructions/Material

Safety Data Sheets.

HSNO Hazard Classification 6.1D; 6.9A; 9.1B; 9.3C **Poisons Schedule (Aust)** No Data Available

3. COMPOSITION / INFORMATION ON INGREDIENTS

Ingredients

Chemical Entity	Formula	CAS Number	Proportion
Manganese (li) Sulfate Monohydrate	No Data Available	10034-96-5	>98.0 %













4. FIRST AID MEASURES

Eyes

Description of necessary measures according to routes of exposure:

Swallowed Rinse mouth with water. If swallowed, give a glass of water to drink. If vomiting occurs give

further water. Never give anything by mouth to an unconscious person. Seek medical assistance.

Immediately flush eyes with water for at least 15 minutes while holding eyelids open. In all cases

of eye contamination, its is a sensible precaution to seek medical advice.

Skin Immediately remove all contaminated clothing and shoes. Flush skin and hair with running water

(and soap if available). Seek medical attention in the event of irritation.

Inhaled Remove victim from exposure to fresh air - avoid becoming a casualty. Remove contaminated

clothing and loosen remaining clothing. Allow patient to assume most comfortable position and keep warm and at rest. If patient finds breathing difficult, and develops a bluish discolouration of the skin, ensure airways are clear, and have qualified person give oxygen through a face mask.

Seek immediate medical advice.

Advice to Doctor Treat symptomatically based on judgement of doctor and individual reactions of patient.

Medical Conditions Aggravated

Chronic: Chronic inhalation or ingestion may result in manganism characterized by neurological symptoms such as headache, apathy, and weakness of the legs, followed by psychosis and neurological symptoms similar to those of Parkinson's disease. In its acute form, manganese poisoning has an effect characteristic of other heavy metals, leading to "metal fume fever" if dust or fume is inhaled in sufficient quantity. An airborne concentration thought to be immediately dangerous to life or health is in the order of 10,000 mg/m³ (as Manganese). May impair fertility. Other chronic effects from inhaling high amounts of manganese include an

increased incidence of cough and bronchitis and susceptibility to infectious lung disease.

Long Term Effects: Available evidence from animal studies indicate that repeated or prolonged exposure to this material could result in effects on the lungs, and central nervous system.

Repeated inhalation may cause chronic bronchitis.

5. FIRE FIGHTING MEASURES

Flammability Conditions Product is a non-flammable solid.

Extinguishing Media If material is involved in a fire use Fine water spray, normal foam, dry agent (carbon dioxide, dry

chemical powder).

Hazardous Products of Combustion Non-combustible solid. Incompatible with strong oxidising agents, strong acids, powdered

metals, hydrogen peroxide, and sources of ignition. When involved in a fire, this product may

generate carbon dioxide, sulphur oxide and sulphur dioxide, and oxides of manganese.

Personal Protective Equipment Fire fighters should wear a positive-pressure self-contained breathing apparatus (SCBA) and

protective fire fighting clothing (includes fire fighting helmet, coat, trousers, boots and gloves) or chemical splash suit. Clear fire area of all non-emergency personnel. Stay upwind. Keep out of low areas. Eliminate ignition sources. Move fire exposed containers from fire area if it can be done without risk. Do NOT allow fire fighting water to reach waterways, drains or sewers. Store

fire fighting water for treatment.

Flash Point

No Data Available
Lower Explosion Limit

No Data Available
Upper Explosion Limit

Auto Ignition Temperature

No Data Available
Hazchem Code

No Data Available













6. ACCIDENTAL RELEASE MEASURES

General Response Procedure

Personnel involved in the clean up should wear full protective clothing as listed in section 8. Avoid accidents, clean up immediately. Evacuate all unnecessary personnel. Increase ventilation. Avoid walking through spilled product as it is slippery when spilt. Stop leak if safe to do so. Isolate the danger area. Do NOT let product reach drains or waterways. If product does enter a waterway, advise the Environmental Protection Authority or your local Waste Management. Use clean, non-sparking tools and equipment. Shut off all possible sources of ignition.

Clean Up Procedures

Contain and sweep/shovel up spills with dust binding material or use an industrial vacuum cleaner. Water may be used to prevent dusting. Transfer to a suitable, labelled container and dispose of promptly as hazardous waste. Wash area down with excess water.

7. HANDLING AND STORAGE

Handling

Ensure an eye bath and safety shower are available and ready for use. Observe good personal hygiene practices and recommended procedures. Wash thoroughly after handling. Take precautionary measures against static discharges by bonding and grounding equipment. Avoid contact with eyes, skin and clothing. Do not inhale product vapours. Avoid prolonged or repeated exposure. Remove contaminated clothing and wash before reuse.

Storage

Store in a cool, dry, well-ventilated area. Keep containers tightly closed when not in use. Inspect regularly for deficiencies such as damage or leaks. Protect against physical damage. Store away from incompatible materials as listed in section 10. Protect from direct sunlight and heat. This product has a UN classification of 3077 and a Dangerous Goods Class 9 (Miscellaneous) according to The Australian Code for the Transport of Dangerous Goods By Road and Rail.

NOTE: This product is subject to special provision AU01 according to The ADG7.

SP No. AU01 Environmentally Hazardous Substances meeting the descriptions of UN 3077 or UN 3082 are not subject to this Code when transported by road or rail in; (a) packagings; (b) IBCs; or (c) any other receptacle not exceeding 500 kg(L).

Container

Container type/packaging must comply with all applicable local legislation. Store in original packaging as approved by manufacturer.

8. EXPOSURE CONTROLS / PROTECTION

General

The following exposure standard has been established by The Australian Safety and Compensation Council (ASCC); Manganese, dust & compounds (as Mn): 8hr TWA = 1 mg/m3 NOTE: The exposure value at the TWA is the average airborne concentration of a particular substance when calculated over a normal 8 hour working day for a 5 day working week. These exposure standards are guides to be used in the control of occupational health hazards. All atmospheric contamination should be kept to as low a level as is workable. These exposure standards should not be used as fine dividing lines between safe and dangerous concentrations of chemicals. They are not a measure of relative toxicity.

No Data Available

Exposure Limits Biological Limits

No information available on biological limit values for this product.













Engineering Measures

A system of local and/or general exhaust is recommended to keep employee exposures as low as possible. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing dispersion of it into the general work area. Long-term exposure at or below 0.5 to 1 mg/m³ should afford protection for those individuals who may be susceptible to the neurological effects of prolonged exposure to manganese. Accordingly, the Exposure Standards Working Group recommends an eight-hour TWA exposure standard of 1 mg/m³ for manganese and inorganic compounds. Although there is acknowledgement that the particle size distribution, type of manganese compound & oxidation state may play an important role in the development of both the neurological and respiratory effects of manganese, there is insufficient evidence to distinguish confidently between manganese fume, dust and other inorganic manganese compounds. For this reason it is recommended that a single exposure standard be applied for all inorganic manganese compounds, measured as inspirable dust (as Mn).

9. PHYSICAL AND CHEMICAL PROPERTIES

Physical State Solid

Appearance Crystalline Powder

 Odour
 Odourless

 Colour
 Grey/Pink

 pH
 3 - 3.5

Vapour PressureNo Data AvailableRelative Vapour DensityNo Data AvailableBoiling/Melting Point850 (Decomposes) °CSolubilitySoluble in water °C

Freezing Point 700 °C Specific Gravity 2.95

Flash Point

Auto Ignition Temp

No Data Available

Evaporation Rate

No Data Available

Bulk Density

Corrosion Rate

No Data Available

No Data Available

Decomposition Temperature 850 °C

Density No Data Available **Specific Heat** No Data Available Molecular Weight No Data Available **Net Propellant Weight** No Data Available **Octanol Water Coefficient** No Data Available **Particle Size** No Data Available **Partition Coefficient** No Data Available **Saturated Vapour Concentration** No Data Available













Vapour Temperature No Data Available **Viscosity** No Data Available **Volatile Percent** No Data Available **VOC Volume** No Data Available **Additional Characteristics** No Data Available **Potential for Dust Explosion** No Data Available **Fast or Intensely Burning Characteristics** No Data Available Flame Propagation or Burning Rate of No Data Available

Solid Materials

Non-Flammables That Could Contribute

Unusual Hazards to a Fire

Properties That May Initiate or Contribute No Data Available

to Fire Intensity

Reactions That Release Gases or Vapours **Release of Invisible Flammable Vapours**

and Gases

No Data Available

No Data Available No Data Available

STABILITY AND REACTIVITY 10.

Chemical Stability Product is stable under normal conditions of use, storage and temperature. Material loses all

water at 400 - 450 deg C.

Avoid excessive heat, direct sunlight, generating dust, moisture, confined spaces. **Conditions to Avoid**

Materials to Avoid Incompatible with strong oxidising agents, strong acids, powdered metals, hydrogen peroxide,

and sources of ignition.

Hazardous Decomposition Products When involved in a fire, this product may generate carbon dioxide, sulphur oxide and sulphur

dioxide, and oxides of manganese.

Hazardous Polymerisation Hazardous polymerisation will not occur. May react violently with hydrogen peroxide. Manganese

sulphate can react with strong acids, strong oxidising agents and powdered metals.

11. **TOXICOLOGICAL INFORMATION**

General Information Oral LD50 Rat: 2150mg/Kg

Oral LD50 Mouse: 2330mg/Kg

Eye Irritant May be an eye irritant. Exposure to the dust may cause discomfort due to particulate nature. May

cause physical irritation to the eyes.

Ingestion Harmful: danger of serious damage to health by prolonged exposure if swallowed. Swallowing

> can result in nausea, vomiting, diarrhoea, and abdominal pain. A side effect of oral manganese administration is an increase in losses of calcium in the faeces and a subsequent lowering of

calcium blood levels.

Inhalation Material may be irritant to the mucous membranes of the respiratory tract (airways).

Harmful: danger of serious damage to health by prolonged exposure through inhalation.

Skin Irritant Contact with skin may result in irritation.

Carcinogen Category













ECOLOGICAL INFORMATION

Ecotoxicity Toxic to aquatic organisms. May cause long term adverse effects in the aquatic environment.

Persistence/Degradability No information available on persistence/degradability for this product. No information available on mobility for this product. Soluble in water Mobility

Environmental Fate Do NOT let product reach waterways, drains and sewers. **Bioaccumulation Potential** No information available on bioaccumulation for this product.

Environmental Impact No Data Available

13. **DISPOSAL CONSIDERATIONS**

General Information Dispose of in accordance with all local, state and federal regulations. All empty packaging should

be disposed of in accordance with Local, State, and Federal Regulations or

recycled/reconditioned at an approved facility.

Special Precautions for Land Fill Contact a specialist disposal company or the local waste regulator for advice.

TRANSPORT INFORMATION 14.

ADG Code Non-Dangerous Goods according to the criteria of the Australian Dangerous Goods

Code (ADG Code).

Air

IATA

ENVIRONMENTALLY HAZARDOUS SUBSTANCE, SOLID, N.O.S. **Proper Shipping Name**

(Manganese Sulphate Monohydrate)

9 Misclaneous Dangerous Goods and Articles Class

Subsidiary Risk(s) No Data Available

3077 **UN Number** Hazchem 2Z **Pack Group** III

No Data Available **Special Provision**

Land

Australia: ADG Code

Proper Shipping Name MANGANESE SULPHATE, MONOHYDRATE

Class No Data Available Subsidiary Risk(s) No Data Available **EPG** No Data Available **UN Number** No Data Available Hazchem No Data Available **Pack Group** No Data Available

Special Provision SPAU01

New Zealand: NZS5433

Proper Shipping Name ENVIRONMENTALLY HAZARDOUS SUBSTANCE, SOLID, N.O.S.

(Manganese Sulphate Monohydrate)

Class 9 Miscllaneous Dangerous Goods and Articles

Subsidiary Risk(s) No Data Available

47 Low To Moderate Hazard Substances **EPG**

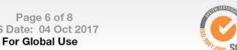
UN Number 3077 Hazchem 27 **Pack Group** III

Special Provision No Data Available













Sea IMDG

Proper Shipping Name ENVIRONMENTALLY HAZARDOUS SUBSTANCE, SOLID, N.O.S.

(Manganese Sulphate Monohydrate)

Class 9 Miscllaneous Dangerous Goods and Articles

Subsidiary Risk(s) No Data Available

 UN Number
 3077

 Hazchem
 2Z

 Pack Group
 III

Special Provision No Data Available

EMS FA,SF Marine Pollutant Yes

15. REGULATORY INFORMATION

General Information No Data Available

EPA (New Zealand) Hazardous Substances and New Organisms Act (HSNO)

Approval Code: HSR004019

Poisons Schedule (Aust) No Data Available

AICS Name SULFURIC ACID, MANGANESE(2+) SALT (1:1) MONOHYDRATE

16. OTHER INFORMATION

Key/Legend < Less Than

> Greater Than

AICS Australian Inventory of Chemical Substances

atm Atmosphere

CAS Chemical Abstracts Service (Registry Number)

cm² Square CentimetresC0₂ Carbon Dioxide

COD Chemical Oxygen Demand

deg C (°C) Degrees Celcius

EPA (New Zealand) Environmental Protection Authority of New Zealand

deg F (°F) Degrees Farenheit

g Grams

g/cm³ Grams per Cubic Centimetre

g/I Grams per Litre

HSNO Hazardous Substance and New Organism

IDLH Immediately Dangerous to Life and Health

immiscible Liquids are insoluable in each other

inHg Inch of Mercury
inH₂0 Inch of Water
K Kelvin
kg Kilogram

kg/m³ Kilograms per Cubic Metre

Ib Pound

LC stands for lethal concentration, LC50 is the concentration of a material

in air which causes the death of 50% (one half) of a group of test animals. The material is inhaled over a set period of time, usually 1 or 4 hours.

LD stands for Lethal Dose. LD50 is the amount of a material, given all at

once, which causes the death of 50% (one half) of a group of test animals.











Itr or L Litre

m³ Cubic Metre mbar Millibar mg Milligram

mg/24H Milligrams per 24 Hours
mg/kg Milligrams per Kilogram
mg/m³ Milligrams per Cubic Metre

Misc or Miscible Liquids form one homogeneous liquid phase regardless of the amount of

either component present.

mm Millimetre

mmH₂0 Millimetres of Water mPa.s Millipascals per Second

N/A Not Applicable

NIOSH National Institute for Occupational Safety and Health
NOHSC National Occupational Heath and Safety Commission
OECD Organisation for Economic Co-operation and Development

Oz Ounce

PEL Permissible Exposure Limit

Pa Pascal

ppb Parts per Billion**ppm** Parts per Million

ppm/2h Parts per Million per 2 Hours
ppm/6h Parts per Million per 6 Hours
psi Pounds per Square Inch

R Rankine

RCP Reciprocal Calculation Procedure
STEL Short Term Exposure Limit
TLV Threshold Limit Value

tne Tonne

torr Millimetre of Mercury
TWA Time Weighted Average
ug/24H Micrograms per 24 Hours

UN United Nations wt Weight

Disclaimer

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Revision: 1

SDS Date: 04 October 2017

End of SDS







